



**TGM for the Implementation of the Hawai'i State Contingency Plan
Section 11.0
HANDLING AND ANALYSIS OF SAMPLES**

**SECTION 11
HANDLING AND ANALYSIS OF SAMPLES**

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Section 11

Acronyms and Abbreviations

SECTION 11 ACRONYMS AND ABBREVIATIONS

°C	Degree Celsius
≤	Less than or equal to
A	Hold time after extraction until analysis
APHA	American Public Health Association
ASTM	American Society for Testing and Materials
BTEX	Benzene, toluene, ethylbenzene, and total xylenes
CFR	Code of Federal Regulations
CWA	Clean Water Act
Dioxins	Polychlorinated dibenzodioxins
DOT	U.S. Department of Transportation
DQO	Data Quality Objective
E	Hold time to extraction
EAL	Environmental Action Level
Furans	Polychlorinated dibenzofurans
HCl	Hydrochloric acid
HDOA	Hawai'i Department of Agriculture
HDOH	Hawai'i Department of Health
HEER Office	Hazard Evaluation and Emergency Response Office
HNO ₃	Nitric acid
HVOCs	Halogenated Volatile Organic Compounds
L	Liter
MIS	Multi Increment sample(s)
mL	Milliliter
mm	Millimeter
MTBE	Methyl tertiary butyl ether
Na ₂ SO ₄	Sodium bisulfate
NaOH	Sodium hydroxide
NaS ₂ O ₃	Sodium thiosulfate
NELAC	National Environmental Laboratory Accreditation Conference
NIOSH	National Institute for Occupational Safety and Health
PAH	Polynuclear Aromatic Hydrocarbon
PCB	Polychlorinated Biphenyl
PPE	Personal Protective Equipment
QA/QC	Quality Assurance and Quality Control
SAP	Sampling and Analysis Plan
SDWA	Safe Drinking Water Act
SOP	Standard Operating Procedure
SVOC	Semi-volatile organic compound
TGM	Technical Guidance Manual
TPH-D	Total Petroleum Hydrocarbons-diesel range
TPH-G	Total Petroleum Hydrocarbons-gasoline range

TPH-O	Total Petroleum Hydrocarbons-oil range
TSA	Transportation Security Administration
US Navy	United States Department of the Navy
USACE	United States Army Corps of Engineers
USDA	United States Department of Agriculture
USEPA	United States Environmental Protection Agency
VOA	Volatile organic analysis
VOC	Volatile Organic Compound



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11.0 HANDLING AND ANALYSIS OF SAMPLES

This section presents appropriate methods for containing, preserving, shipping, and analyzing environmental samples for projects conducted under the guidance of this Technical Guidance Manual (TGM). Proper handling and analysis of environmental samples is an essential step to collecting representative, defensible data and to achieving Data Quality Objectives (DQO). The Sampling and Analysis Plan (SAP) documents necessary sample handling and analysis procedures and should identify the matrix to be sampled, the type and volume of proposed sample containers, preservatives (if any), and the proposed analytical methods for each sampling event.

This section focuses on soil and groundwater samples collected as part of an environmental investigation, which are sent to a fixed laboratory for analysis. Mobile laboratories used for field analysis may also be employed during an environmental investigation, but the use of mobile laboratories should be carefully planned and documented in the SAP prior to field activities and analysis. The State of Hawai'i Department of Health (HDOH) Hazard Evaluation and Emergency Response Office (HEER Office) recommends the submittal of SAPs proposing the use of mobile laboratories for review prior to the initiation of sample collection activities.

During the collection, handling, and packing activities, personnel should use proper personal protective equipment (PPE). Typical PPE for use in sample collection, handling, and packing should include gloves and safety glasses, especially when employing chemical preservatives. Site-specific conditions may require the use of additional PPE (e.g., chemical protection clothing) based on the hazards present at the site, which should be documented in the site-specific Safety and Health Plan within the SAP.

Packing extra sample containers is recommended when planning fieldwork, in case additional sampling locations or areas are identified during execution of the fieldwork. Breakage, both prior to and after sample collection should be considered, particularly in light of challenges presented by remote field sites, as well as shipping requirements unique to performing investigations in Hawai'i.

Guidance is presented on the use of soil sampling equipment in [Section 5](#) and on the use of groundwater or surface water sampling equipment in [Section 6](#). A discussion of gas phase samples is not included in this section; guidance on handling and analysis of soil vapor or indoor air samples is presented in [Section 7](#). Quality assurance/quality control (QA/QC) considerations, such as trip blanks and equipment rinsate blanks, are discussed in [Section 10](#).



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Section 11.1

SAMPLE CONTAINERS

11.1 SAMPLE CONTAINERS

Recommended sample containers for soil are presented in [Table 11-A](#) in the [Appendix 11-A](#), while [Table 11-B](#) in the Appendix 11-B present the recommended sample containers for groundwater. The sample volumes for the containers listed in the tables represent the recommended size of container. More than one sample container may be required for the associated parameter and method. If soil or groundwater samples are to be analyzed for multiple contaminants, the sample volumes listed in the tables will require adjustment. Always consult with the laboratory when planning fieldwork to ensure that the proper sample containers and preservatives are used and sufficient sample mass/volume is collected for all intended analyses.

11.1.1 SOIL SAMPLE CONTAINERS

The types of sample containers used for the collection of soil samples is dependent upon the characteristics of media to be sampled as well as the specific analysis to be performed. Other factors, such as the anticipated concentrations of contaminants and the desired reporting limits are also important to consider when selecting appropriate soil sample containers.

In general, soil sample containers may be grouped into the following three broad categories: (1) non-volatile soil sample containers, (2) volatile soil sample containers, and (3) *Multi-Increment* soil sample containers (note that *Multi-Increment* samples may include both non-volatile and volatile analysis).

11.1.1.1 NON-VOLATILE SOIL SAMPLE CONTAINERS

Soil samples for non-volatile analysis are typically collected in wide-mouth glass jars sealed with Teflon-lined caps. Various volumes of wide-mouth glass jars, ranging from 2 ounces to 16 ounces, are employed in soil sample collection and are available as certified pre-cleaned prior to shipment to the end user. During sample collection, the soil is transferred directly from the sampling device (e.g., split-spoon sampler or acrylic tube liner) to the glass jar, which is sealed with a Teflon-lined cap.

Soil samples for non-volatile analysis may also be collected in stainless steel or brass tubes sealed with Teflon tape and plastic end caps, or in some cases in plastic bags. The stainless steel or brass tubes are typically 6 inches in length and vary between 1.5 and 3 inches or greater in diameter. The tubes are typically placed as a liner in a split spoon sampler driven by a drill rig. During sample collection, the soil is retained in the tubes following retrieval from the sampler, and Teflon tape and a plastic end cap is immediately placed over each end of the tube.

11.1.1.2 VOLATILE SOIL SAMPLE CONTAINERS

United States Environmental Protection Agency (USEPA) Method 5035 describes a closed-system purge-and-trap process for the analysis of volatile organic compounds in solid materials (e.g., soils, sediments, and solid waste) ([USEPA, 1997g](#)), which was subsequently updated with Method 5035A ([USEPA, 2002h](#)). The method is designed for use on samples containing low levels of VOCs, but procedures are also provided for collecting and preparing solid samples containing high concentrations of VOCs and for oily wastes. The procedures in EPA Method 5035 and 5035A may be used in conjunction with any appropriate determinative gas chromatographic procedure, including, but not limited to, EPA Methods 8015, 8021, and 8260.

Soil samples for volatile analysis can be collected using an EnCore® sampler, TerraCore® sampler, or similar sampling device capable of collecting a known mass of soil, approximately 5 grams, for preservation and analysis. The soil aliquot in the sampler may be chemically preserved by placing it into a glass jar containing a known quantity of preservative (see [Subsection 11.2.3](#) for additional discussion of volatile soil sample collection).

The use of glass jars without the use of chemical preservatives when collecting soil samples for volatile analysis is not recommended.

The use of stainless steel or brass tubes when collecting soil samples for volatile analysis is not recommended.

11.1.1.3 MULTI-INCREMENT SOIL SAMPLE CONTAINERS

The collection of a 30- to 50-increment sample can result in approximately 500 to 2,000 grams of soil (i.e., the *Multi-Increment* sample) depending upon the mass of each increment. The sample container for a *Multi-Increment* sample must be large enough to accommodate this mass. For non-volatile analysis, a common method for collecting a *Multi-Increment* bulk sample is to place each increment into a dedicated, disposal plastic bag (such as a 2 gallon zip top or a heavy duty trash compactor bag) as each

increment is collected. The bag may be sealed, labeled and submitted directly to the laboratory for sub-sampling and analysis. There is a potential for some semi-volatile organic compounds (SVOCs), in particular, phthalates, to be transferred from plastic bags to samples, particularly if the soil has coarse or sharp particles that may abrade the bag surface. Therefore, when collecting *Multi-Increment* samples, the sampling team should consider the use of clean wide-mouth glass jars to collect and/or transport samples to be analyzed for SVOCs. These types of jars are available commercially and from Hawai'i-based analytical laboratories.

As discussed in [Section 4.2.1](#), the *Multi-Increment* sample for non-volatile analysis may be sub-sampled in the field if an appropriate Standard Operating Procedure (SOP) has been developed for field sub-sampling and included in the SAP. In this case, the *Multi-Increment* sample would be collected in a dedicated, disposable plastic bag or wide-mouth jar, which is transferred to a pan for sieving and sub-sampling. The sub-sample is collected into a plastic zip top bag or glass jar and submitted to the laboratory for analysis. Field sub-sampling should follow procedures similar to those employed by the analytical laboratory as described in [Section 4.2.2](#). Field sub-sampling of the *Multi-Increment* sample to collect a representative sub-sample may be difficult to perform due to conditions commonly encountered at field sites in Hawai'i (e.g., trade winds and/or rain). Laboratory sub-sampling to collect a representative sub-sample is generally preferable due to the controlled environmental conditions. Whenever samples are sub-sampled in the field, fully document the procedures, equipment, training requirements, and QA/QC measures employed during field sub-sampling in the SAP.

For laboratory sub-sampling, the most common QA/QC measure includes the collection of sub-sampling replicates (typically triplicates) to assess the error introduced by the sub-sampling routine. See discussion of replicates for laboratory sub-sampling in [Section 10.7.6](#).

The recommended approach for the collection of *Multi-Increment* soil samples intended for volatile analysis is discussed in [Section 4.2.7](#).

11.1.2 WATER SAMPLE CONTAINERS

The type of sample container used for collecting surface water or groundwater samples is dependent upon the specific analysis to be performed. Other factors, such as the anticipated concentrations of contaminants, the desired reporting limits, and the presence of free product are also important to consider when selecting appropriate water sample containers.

In general, water sample containers may be grouped into the following two broad categories: (1) non-volatile water sample containers, and (2) volatile water sample containers.

11.1.2.1 NON-VOLATILE WATER SAMPLE CONTAINERS

Water samples for organic non-volatile analysis are typically collected in 1 liter amber glass jars without the use of chemical preservatives. When collecting groundwater samples, fill the water sample containers by directing the outlet of the sampling device (i.e., pump tubing or bailer) toward the top and side of the sample container to allow the water to run down the inside of the bottle. Avoid agitation and the creation of bubbles when collecting water samples. To prevent cross contamination, avoid contacting the interior or top of the sample containers with either the sampling device or gloved hands.

Water samples for dissolved metals analysis are collected in 250 milliliter (mL) plastic bottles and preserved with nitric acid to a pH less than 2. Dissolved metals analysis also requires filtration prior to collection as discussed in [Section 6](#).

11.1.2.2 VOLATILE WATER SAMPLE CONTAINERS

Water samples for volatile analysis are typically collected in 40 mL glass jars with septum-sealed screw caps and preserved with hydrochloric acid (HCl) to a pH less than 2. During sample collection, the screw caps should be carefully placed on the jars and sealed with zero headspace. When collecting groundwater samples for volatile analysis, fill the 40 mL jars by directing the outlet of the sampling device toward the top and side of the sample container to allow the water to run down the inside of the bottle. Adjust the flow rate of the water sampling device so that it does not cause the jars to rapidly overflow, causing loss of VOCs, sample volume, or any sample preservative. Avoid agitation and the creation of bubbles when collecting volatile water samples to prevent the loss of volatile constituents. To prevent cross contamination, avoid contacting the interior or top of the sample containers with either the sampling device or gloved hands.

The preservative HCl may react (effervesce) with turbid water containing calcareous particulates, resulting in a loss of volatile constituents. If a strong reaction is observed when filling the 40 mL jars containing HCl, collect the water samples for volatile analysis as unpreserved samples and note on the chain-of-custody. The collection of unpreserved water samples reduces the hold time of 14 days down to 7 days.



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Section 11.2

SAMPLE PRESERVATION AND HOLD TIMES

11.2 SAMPLE PRESERVATION AND HOLD TIMES

This section presents the recommended sample preservation for soil and water sample collection. Always consult with the laboratory when planning fieldwork to ensure that the proper sample containers and preservatives are used. [Tables 11-A](#) and [11-B](#), provided in the Appendices, present the recommended preservation and hold times for soil and groundwater, respectively.

Sample preservation consists of methods to assure the samples analyzed in the laboratory are representative of the field conditions. Preservation methods may include maintaining sample temperatures, analyzing the samples within recommended hold times or using chemicals (such as HCl or nitric acid [HNO₃]) to stabilize the target contaminants by altering the sample chemistry.

Several potential difficulties associated with field preservation techniques include: sampling gravely soil matrix (if using EnCore® type samplers), spillage or evaporation of pre-weighed chemical preservatives, shipping restrictions involving chemical preservatives, or potential chemical reactions between calcareous particles and HCl preservative.

11.2.1 SAMPLE TEMPERATURE

Upon collection and sealing of sample containers, immediately begin the cooling process in the field by placing the sample containers in an insulated cooler containing water ice or frozen gel packs. Use of water ice is generally considered by the HEER Office to be more efficient to rapidly cool samples, and may be especially important for use with samples for volatile analyses, when feasible. Maintain the temperature of the sample containers at less than or equal to 6 degrees Celsius (°C) from the time of collection through the delivery to the analytical laboratory.

The National Environmental Laboratory Accreditation Conference (NELAC) states that samples which require thermal preservation shall be considered acceptable if the arrival temperature is within 2°C of the required temperature or the method specified range. For samples with a temperature requirement of 4°C, an arrival temperature from 0°C and 6°C meets specifications. Samples that are delivered to the laboratory on the same day that they are collected may not meet these criteria. In these cases, the samples are considered acceptable if there is evidence that the chilling process has begun, such as arrival at the analytical laboratory on ice.

11.2.2 CHEMICAL PRESERVATION

When employing chemical preservation, add the chemical reagents to the sample containers prior to mobilizing to the field, or request sample containers with pre-measured chemical reagents from the analytical laboratory. Carefully place the soil or groundwater samples into the sample containers to minimize loss of chemical preservative as well as volatile constituents in the sample (i.e., do not overfill water sample containers or leave the cap off a jar containing methanol for soil samples).

Some commonly used preservation chemicals may react with the sample media. For example, calcareous soil may react with sodium bisulfate; turbid groundwater collected from a coral aquifer formation may react with hydrochloric acid. If the sample media reacts with the chemical preservative, volatile organic constituents may be lost due to effervescence during sample collection, so an alternative preservation method or no chemical preservation may be needed.

The use of chemical preservatives for fieldwork in Hawai'i presents challenges when shipping preserved samples between islands or to the mainland (see [Subsection 11.4](#) for additional details on shipping preservatives).

Some samples collected for specific analysis, such as dissolved metals in groundwater, may require pre-treatment prior to collection, as well as preservation (see [Section 6.0](#)).

11.2.3 SAMPLE HOLD TIMES

The soil and water samples should be analyzed as soon as possible after collection. Hold times are the maximum allowable times that a sample, or any subsample generated from the bulk sample, may be held before analysis. Several methods provide guidelines for both the hold time until extraction (denoted by "E" on Tables 11-A and 11-B in the Appendices) and the hold time after extraction until analysis (denoted by an "A" on the Tables).

Results from samples analyzed past the hold times may or may not be usable, depending upon the DQO set forth in the SAP. At a minimum, the results from analyses conducted past the recommending hold times should be interpreted as minimum concentrations.

Volatile Soil Samples

The sample collection and preparation methods in EPA Method 5035 and 5035A include short hold times, ranging from 48 hours up to 14 days (if samples are chemically preserved in the field at the time of collection). The following presents the preservation and associated hold times for volatile soil samples:

- Unpreserved soil samples collected with EnCore® samplers (or equivalent device) and stored at 4°C must be analyzed within 48 hours from the time of sample collection. Samplers are transported to the laboratory where the cap is opened and the undisturbed sample is immediately placed in the appropriate preservation fluid prior to analysis.
- Unpreserved soil samples collected with EnCore® samplers (or equivalent device) and immediately frozen on dry ice in the field to less than negative 7°C may be analyzed within 7 days from the time of sample collection. EPA Region IX has approved, on a limited basis, this modification of the method allowing for the freezing of samples for up to 7 days from sampling date to analysis ([USEPA, 2005b](#)).
- Unpreserved soil samples collected with EnCore® samplers (or other similar sampling device capable of providing a pre-determined mass of soil) and immediately extruded into a glass jar containing laboratory grade water must be analyzed within 14 days from the time of sample collection only if the samples are delivered to the laboratory and frozen within 48 hours of collection. Sample container breakage may be a concern when using this approach since the glass jars are not intended to contain freezing liquids. If employing this approach, consider collecting additional sample volumes at each sample location to account for breakage.
- Preserved soil samples collected with EnCore® samplers (or other similar sampling device capable of providing a pre-determined mass of soil) and immediately extruded into a glass jar containing sodium bisulfate preservative must be analyzed within 14 days from the time of sample collection. This approach is appropriate for non-calcareous soils and a field check to determine whether the soil is calcareous is recommended (calcareous formations are commonly encountered when performing fieldwork in Hawai'i). The use of sodium bisulfate as a preservative should be discussed with the analytical laboratory since it may cause laboratory instrument problems requiring frequent recalibration.
- Preserved soil samples collected with EnCore® samplers (or other similar sampling device capable of providing a pre-determined mass of soil) and immediately extruded into a glass jar containing methanol must be analyzed within 14 days from the time of sample collection. This approach is appropriate if high levels of contaminants are suspected, as well as for any contaminants for which lab reporting limits below applicable HEER Office Environmental Action Levels (EALs) can be achieved ([HDOH, 2008](#)). In this application, methanol acts as the preservative as well as the extraction solvent and a limited volume of methanol is introduced directly into the laboratory instrument. As such, the dilution and the reporting limits are higher than with other preservation and extraction approaches. In some cases, the laboratory may be able achieve lower reporting limits for specific contaminants preserved in methanol by using additional analytical techniques, so consultation with the laboratory is recommended.



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Section 11.3

SAMPLE CONTROL AND CHAIN-OF-CUSTODY PROCEDURES

11.3 SAMPLE CONTROL AND CHAIN-OF-CUSTODY PROCEDURES

Sample control and chain-of-custody procedures are extremely important for establishing that sample integrity was maintained from the time of collection through the time of analysis. Sample control procedures include the use of unique sample identifications, and sample labeling requirements. Chain-of-custody procedures include the use of chain-of-custody forms and custody seals.

11.3.1 SAMPLE IDENTIFICATION AND LABELS

Each sample collected in the field must be provided with a unique sample identification. In general, sample identification may include some or all of the following information:

- Project number (if samples collected by a consultant)
- Project location
- Sample location information (i.e., borehole or monitoring well identification)
- Depth of sample collection (for subsurface soil samples)
- Date/Time reference

A date/time reference is recommended if multiple samples are anticipated to be collected over the course of the project (such as during long-term monitoring projects).

Each sample collected in the field must be properly labeled using laboratory supplied (or equivalent) labels completed with indelible ink. The sample labels should contain the following information:

1. Sample identification number
2. Sampling date
3. Sampling time
4. Type of preservation
5. Analysis requested
6. Initials of sampler

Additional sample information should be documented in the field log including, but not limited to, the following:

- sample collection method (manual sampling, direct push drill rig)
- type of sample (e.g. *Multi-Increment* or discrete)
- significant observations noted during sample collection (petroleum odor or staining in soil, petroleum product or sheen on water surface)
- a cross reference of primary and replicate QA/QC samples

The peel and stick sample label should be securely affixed to the sample container. The outside of the sample container should be thoroughly cleaned and dried prior to affixing the label. Application of clear plastic adhesive tape over the label on the sample container provides a secondary means of securing the label to the container. Use of an indelible pen to mark the sample identification number on the container is a good backup method that can be used to identify a sample container in the event that it gets separated from the label.

Do not identify or cross-reference QA/QC samples on the sample labels or on the chain-of-custody form (see [Section 10](#) for additional guidance on QA/QC samples).

11.3.2 CHAIN-OF-CUSTODY

Chain-of-custody is the process by which authorized custody of a sample is successively transferred from one person to another by the use of approved procedures and documents. If sample integrity is to be defensible, chain-of-custody procedures are necessary to document handling of samples from procurement through final analysis and disposal.

A sample is considered to be under a person's custody if:

- The sample is in the person's physical possession.
- The sample is in view of the person after that person has taken possession.
- The sample is secured by that person so that no one can tamper with the sample.
- The sample is secured by that person in an area where access is restricted to authorized personnel.

11.3.2.1 CHAIN-OF-CUSTODY FORMS

Chain-of-custody forms are used for tracking the samples from the time of collection in the field through the time of analysis. The chain-of-custody form contains the following information:

- Project identification
- Sampler's name
- Sender - Company name and address
- Destination - Laboratory name and address
- Sample identification
- Number of sample containers per sample
- Preservation, if any
- Date and time of sample collection for each sample
- Requested analyses
- Special handling requirements, if any
- Shipping company
- Printed name and signature of person relinquishing custody, and date and time when custody relinquished
- Printed name and signature of person receiving custody, and date and time when custody received.

Complete chain-of-custody forms at the time of sample collection and prior to leaving the field site. Analytical laboratories typically provide company-specific chain-of-custody forms, and sample labels, if sample containers are procured through the analytical laboratory. Complete chain-of-custody forms with indelible ink.

When transferring samples, the individuals involved must sign, date, and record the time in the relinquished/received-section on the form. The sampler retains one copy of the chain-of-custody form when relinquishing the samples following sample collection. Completely fill out all applicable sections of the form and numerically sequence the forms (i.e., page 1 of 3, etc.) if more than one chain-of-custody form is used for a sample batch. Consider grouping similar sample media in chain-of-custody batches for submittal to the laboratory (i.e., submit groundwater samples on one chain-of-custody batch separate from soil samples collected for the same project).

Chain-of-Custody					Control Number: 94H0	
GTO/DO Manager:			BM To:		Date Page of	
GTO/DO Name:			Company:		Sample Disposal	
GTO/DO Number:			Address:		Shipment	
Default results to the address above or as stated in contract					Comments:	
Cooler No:			# of containers:			
QC Level:			TAT:		Preserve/Freeze:	
Sample Date			Matrix/QC			
Sample ID (COC ID)	Sample ID (New/TP Use Only)	Date Collected	Time Collected	Lab ID	Soil	Water
					Other (dum, sludge, etc.)	Field Duplicate (MS/MED)
					TPH 8015B	CLP VOAa
					CLP VOAa	CLP SVOAa
					CLP Pesticides	CLP Metals
					EPA 8080 (PCBa only)	EPA 8240
					EPA 8270	Total Lead by EPA 6010
						Extra Volume
						MS/MED
						HOLD
						Total # of Containers
Sampler Signature			Date	Time	For Lab Use	
Relinquished By:			Date	Time	Lab No.:	
Received By:			Date	Time	Does COC match sampler: Y or N	
Relinquished By:			Date	Time	Seals on container: Y or N	
Received By (LAB):			Date	Time	Received within holding time: Y or N	
					COC and Inset: Y or N	
					Any other problems: Y or N	
					If problems, Object contacted: Y or N	
					Date contacted: / /	
					Temperature (°C):	

Original (white), Lab Copy (yellow), Field Copy (pink)

Figure 11-1. Example Chain-of-Custody Form.

[Source: [US Navy, 2007](#)]

Shipping companies (e.g. Federal Express, DHL, etc.) are not expected to sign the chain-of-custody form. However, complete shipping documentation, including tracking numbers, becomes part of the chain-of-custody record. When using shipping companies, the last person to have custody of the samples must fill in both the relinquished section (as normal) as well as the received section (identifying the shipping company's name and the date and time when the shipment was given into custody of the shipping company).

Once received at the laboratory, custody procedures shall apply. It is then the laboratory's responsibility to maintain custody records throughout sample preparation and analysis. An example generic chain-of-custody form is presented in Figure 11-1.

11.3.2.2 CHAIN-OF-CUSTODY SEALS

To ensure sample integrity when shipping samples, use custody seals. The custody seals must be dated and initialed by the personnel responsible for custody of the samples. Intact custody seals when the samples are logged into the laboratory indicate the physical integrity of the sample was not compromised during sample shipment.

Use custody seals on the outside of each container or cooler when shipping samples. Two seals are required, one at the front or opening edge of the cooler and one at the rear or hinged edge of the cooler. Clear tape can be applied over the custody seals when sealing the cooler or container for shipment.

Shipping sample coolers from the outer islands to O`ahu or to mainland laboratories may require inspection by the Transportation Security Administration (TSA), which may break the chain-of-custody if custody seals are only used on the outside of shipping coolers. In this event, ensure individual sample integrity through the use of custody seals on individual sample container lids.

Custody seals may not be necessary if the sampling personnel retain custody of the samples from the time of collection to the time of delivery to the analytical lab, if the samples are delivered in person and a shipping company is not used.



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Section 11.4

SAMPLE SHIPPING

11.4 SAMPLE SHIPPING

Upon completion of sample collection, shipment of samples to an analytical laboratory is typically required. This may include transport from an outer island to an O'ahu-based laboratory or transport to a mainland laboratory. Properly pack the shipping containers to protect the individual sample containers, maintain the samples at temperature, and comply with all applicable transportation regulations.

Chain-of-custody forms are typically placed inside a sealed bag and adhered to the interior of the shipping container. The shipping paperwork is adhered to the outside of the shipping container and the container sealed for shipment. Shipping paperwork need only be attached to one cooler if multiple coolers are used in a single shipment, provided the shipping company concurs with this practice. When shipping multiple coolers in a single shipment, use additional labels to identify the cooler along with the total number of coolers in the shipment (e.g., cooler 2 of 3).

11.4.1 SAMPLE PACKING

Standard coolers are typically employed when packing samples for shipment although other shipping containers are acceptable. Interior packing materials include bubble wrap or foam sleeves encasing the individual sample containers and lining the cooler interior. Interior packing materials should be sufficient to prevent breakage during transport. After placing individually wrapped sample containers in the cooler, fill all empty space between sample containers with padding to minimize movement against each other.

Containers for water samples should be packed in an upright position and not stacked on their sides. When shipping water samples, line the bottom of the cooler with absorbent material to contain liquids in case of breakage.

The use of frozen gel ice is convenient to maintain the samples at a temperature of less than 6°C, but water ice may also be used. When using water ice, precautions to prevent spillage from the sample cooler (i.e., containing water ice in triple bags and sealing the cooler drain) are essential to expedient delivery of the shipment to the laboratory. Line the bottom of the cooler with absorbent material when using water ice in coolers.

Immediately prior to shipment, replace the ice or frozen gel packs in the coolers so that samples will be maintained at less than 6°C during transport to the analytical laboratory.

11.4.2 SHIPPING OR DELIVERY

If utilizing a mainland laboratory, consider shipping and delivery constraints resulting from Hawai'i's geographic location, especially when collecting samples with a short hold time. Ensure sample delivery to the analytical laboratory so that there is sufficient time for analysis of the constituent with the shortest hold time. Samples preserved at less than 6°C using gel packs or ice should be shipped via overnight delivery. If samples are shipped on Friday, arrange for Saturday delivery with the analytical laboratory to ensure the correct temperature is maintained throughout shipping.

Errors in shipping and delivery risk exceeding either the required sample temperatures or analysis hold times. As noted previously, results from samples exceeding the required preservation temperatures or analyzed past the hold times may or may not be usable, depending upon the DQO set forth in the SAP. If the data are deemed usable, the results from the analyses should be interpreted as minimum concentrations.

When shipping air, soil, or water samples either inter-island or to the mainland, follow all appropriate U.S. Department of Transportation (DOT) regulations, specifically Title 49 of the Code of Federal Regulations (CFR), Parts 171 through 180 ([Title 49 CFR](#)). The DOT rules and regulations apply to all samples shipped including methanol, sodium bisulfate, and nitric acid preserved samples if a commercial carrier such as FedEx transports the samples. Personnel responsible for shipping soil and water samples in Hawai'i should receive training and refresher training in these regulations. Specific packaging, labeling, and documentation are required by DOT regulations for most samples containing chemical preservatives. In addition, shipping dry ice is also subject to DOT regulations including, but not limited to, shipping container labeling requirements and restrictions on the weight of dry ice included in each shipping container.

Shipment of soil samples from Hawai'i to the mainland is also subject to United States Department of Agriculture (USDA) inspection and regulation. The USDA does not need to inspect water sample shipments. A "USDA Soil Import Permit" is

required to prove that the receiving analytical laboratory is certified by the USDA to receive and properly dispose of soil. In addition, all soil sample coolers must be inspected by a USDA representative, affixed with a label indicating that the coolers contain environmental samples, and accompanied by shipping forms stamped by the USDA inspector prior to shipment ([US Navy, 2007](#)).

In Hawai'i, soil sample shipments are typically brought to the shipping company at the airport where the shipping company contacts a USDA representative to request an inspection. Alternatively, individuals or consulting firms may enter into an agreement with the USDA to ship soil samples. In this way, the USDA does not need to inspect each soil sample shipment. Consider USDA inspection requirements when planning sample shipments and employ custody seals on each individual sample container to ensure proper chain-of-custody control in the event coolers are opened by the USDA for inspection.

Inter-island shipment of soil in Hawai'i is subject to inspection and regulation by the State of Hawai'i Department of Agriculture (HDOA). Annual intra-state permits for shipping soil may be requested from the HDOA Plant Quarantine Branch in Honolulu.

Non-Hazardous Materials Shipment

Field personnel must state whether any sample is suspected to be a hazardous material. Samples may be shipped as non-hazardous based on previous site sample results, field screening results, or visual observations. In addition, environmental samples are currently exempt from Hazardous Goods regulations. Title 40 CFR, Part 261.40(d) states "A sample of solid waste or a sample of water, soil, or air which is collected for the sole purpose of testing to determine its characteristics or composition is not subject to this Part or Parts 262 through 267 or Part 124 of this chapter or to the notification requirements of Section 3010 of RCRA." ([Title 40 CFR](#)) Therefore, no special regulations are required to be followed for the shipment of environmental samples from the field. Note that this provision applies to unpreserved soil and water samples (i.e., no chemical preservatives added during collection).

For groundwater samples specifically, very small quantities of certain dangerous goods may be transported without certain marking and documentation requirements as described in Title 49 CFR Part 172 ([Title 49 CFR](#)). The Hazardous Materials regulations do not apply to the commonly utilized sample preservatives methanol, sodium bisulfate, hydrochloric acid, nitric acid, sulfuric acid, and sodium hydroxide added to water samples if their pH or percentage by weight criteria is met ([USEPA, 1996](#); [USACE, 1998b](#); [US Navy, 2007](#)). Standard preservative volumes in standard sample containers (e.g., HCl in 40 ml volatile organic analysis [VOA] jar) fall under this definition ([USEPA, 1996](#); [USACE, 1998b](#); [US Navy, 2007](#)).

It is extremely important to be aware that regulations may prohibit the shipment of chemically preserved soil samples (such as methanol or sodium bisulfate preserved soil samples intended for volatile analysis). The above paragraph references the shipment of chemical preservatives in water samples only.



TGM for the Implementation of the Hawai'i State Contingency Plan

Section 11.5

APPROVED ANALYTICAL METHODS

11.5 APPROVED ANALYTICAL METHODS

Use the methods and standard operating procedures listed in EPA SW-846 ([USEPA, 1991c](#) and [2003b](#)) for the analyses conducted under the guidance of this TGM. Consider the analytes of interest, the sample matrices, and the minimum detectable concentrations required to accomplish project DQO when selecting analytical methods. [Tables 11-A](#) and [11-B](#) in the Appendices provide the recommended analytical methods for soil and groundwater analysis, respectively.

Use other EPA-approved methods (such as Methods for Chemical Analysis of Water and Wastes ([USEPA, 1983](#)) for analyses that measure parameters such as pH, specific conductance, dissolved oxygen, and temperature. Document any deviations from EPA-approved methods in the SAP.

When EPA-approved methods are not available or appropriate for project-specific requirements, other recognized standard analytical methods, such as those published by the American Society for Testing and Materials (ASTM) International or the National Institute for Occupational Safety and Health (NIOSH), may be used. Example guidance documents for other methods and procedures that may be proposed for site investigations include:

- American Public Health Association (APHA), American Water Works Association, Water Environment Federation. 2005. "Standard Methods for the Examination of Water and Wastewater." 21st Edition ([APHA, 2005](#)).
- ASTM. (updated yearly). "Annual Book of Standards." West Conshohocken, Pennsylvania (ASTM).
- NIOSH. 1994. NIOSH Manual of Analytical Methods, Fourth Edition. Publication No. 94-113 ([NIOSH, 1994](#)).

The published methods are updated at various time intervals. Unless otherwise stated, laboratories conducting work under the guidance of this TGM will use the most current version of any specified analytical method.

On occasion, project-specific conditions might require the use of analytical methods that are either a modification of an EPA-approved method or are not an EPA-approved method. These methods will typically be provided by the laboratory performing the analysis. Any laboratory using modified EPA-approved methods or non-EPA-approved methods must provide a detailed description of sample preparation, instrument calibration, sample analyses, method sensitivity, associated QA/QC requirements, and acceptance criteria, preferably during the planning phases and the creation of the SAP. The laboratory or method developer must provide method performance study information (e.g. detection, recovery, calibration data) to confirm the performance of the method for each applicable matrix. If previous performance studies are not available, they must be developed during the project and included as part of the project results.

An example of a modification of an EPA-approved method that is important for analysis of most soil and sediment samples for metals (whether *Multi-Increment* samples or discrete samples) is EPA Method 3050 for preparation of metals analyses. Soil or sediment samples are typically sieved to the <2 millimeter (mm) particle size before analysis, and an analysis of particles of this size requires a minimum extraction and analysis mass of 10 grams (rather than 1 gram generally recommended in Method 3050) to reduce fundamental error in the analysis (see discussion in [Section 4.2.2](#) and [USEPA, 2003b](#); [ASTM, 2003](#)). Laboratories conducting metals analyses of soils should therefore ensure they have conducted and documented method performance data for the extraction and analysis of these larger masses of soil or sediment.



TGM for the Implementation of the Hawai'i State Contingency Plan Section 11.6 FIELD-BASED MEASUREMENT TECHNOLOGIES

11.6 FIELD-BASED MEASUREMENT TECHNOLOGIES

Field-based measurement technologies have advanced significantly in recent years in terms of quality and utility. Field-based measurement technologies include a broad range of options including devices capable of in-situ measurements, devices capable of providing ex-situ measurements on sampled media in the field, and mobile laboratories that can be transported to the field site.

Some field methods provide qualitative or semi-quantitative analyses that may be used as screening data. A growing number of field techniques, such as field assays and field colorimetric tests, are now available that are capable of providing quantitative, analyte-specific analyses typically associated with standard fixed-laboratory techniques.

Although the HEER Office supports the use of field measurement technologies whenever available and appropriate to enhance or help speed site investigations, these technologies must be augmented by the analysis of a subset of the field samples in a standard fixed analytical laboratory. The results from samples analyzed in the field should be correlated to the results from replicate samples analyzed in the fixed laboratory to assess the precision and accuracy of the field data.

Guidance on field assays and field screening methods is presented in [Section 8](#).



TGM for the Implementation of the Hawai'i State Contingency Plan

Section 11

REFERENCES

SECTION 11 REFERENCES

Citation references for this TGM Section denoted with an "a", "b", "c", etc... after the year of publication may not appear in sequence as these refer to the order placed in the [Master References](#) List for the entire TGM.

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TGM for the Implementation of the Hawai'i State Contingency Plan

Section 11

APPENDICES

SECTION 11 APPENDICES

11-A Laboratory Methods, Containers, Preservation, and Holding Times for Soil Samples

11-B Laboratory Methods, Containers, Preservation, and Holding Times for Groundwater Samples



TGM for the Implementation of the Hawai'i State Contingency Plan
Appendix 11-A
Laboratory Methods for Soil Samples

Appendix 11-A

Laboratory Methods, Containers, Preservation, and Holding Times for Soil Sample

Parameter	Method	Container	Preservation	Holding Time
TPH-G	5035/8015 LUFT	D: EnCore® sampler or equivalent	≤6°C	48 hours
		D: EnCore® sampler or equivalent	Immediately freeze with dry ice to below 0°C	7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Organic-free reagent water and ≤6°C	Laboratory freeze within 48 hours; analysis within 7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Na ₂ SO ₄ and ≤6°C	14 days
		D: 5g soil plug added to 40 mL jar with septum cap	Methanol and ≤6°C	14 days
		MI: 1 L Glass Jar(*)	Methanol and ≤6°C	14 days
TPH-D	3550/8015 3550/8270 3540/8270 LUFT	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days A: 40 days
		D: Stainless Steel or Brass Sleeve		
		MIS: 1 L Glass Jar(*)		
TPH-O	3550/8015B 8440	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days A: 40 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag or Glass Jar		
BTEX, MTBE	5035/8015 5035/8021 5035/8260	D: EnCore® sampler or equivalent	≤6°C	48 hours
		D: EnCore® sampler or equivalent	Immediately freeze with dry ice to below 0°C	7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Organic-free reagent water and ≤6°C	Laboratory freeze within 48 hours; analysis within 7 days

Laboratory Methods, Containers, Preservation, and Holding Times for Soil Sample

Parameter	Method	Container	Preservation	Holding Time
		D: 5g soil plug added to 40 mL jar with septum cap	Na ₂ SO ₄ and ≤6°C	14 days
		D: 5g soil plug added to 40 mL jar with septum cap	Methanol and ≤6°C	14 days
		MI: 1 L Glass Jar(*)	Methanol and ≤6°C	14 days
PAHs, SVOCs	3540/8310 3550/8310 3540/8270C 3550/8270C 8100	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days A: 40 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag or Glass Jar		
HVOCs	8021	D: EnCore® sampler or equivalent	≤6°C	48 hours
		D: EnCore® sampler or equivalent	Immediately freeze with dry ice to below 0°C	7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Organic-free reagent water and ≤6°C	Laboratory freeze within 48 hours; analysis within 7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Na ₂ SO ₄ and ≤6°C	14 days
		D: 5g soil plug added to 40 mL jar with septum cap	Methanol and ≤6°C	14 days
		MI: 1 L Glass Jar(*)	Methanol and ≤6°C	14 days
VOCs	5035/8260	D: EnCore® sampler or equivalent	≤6°C	48 hours
		D: EnCore® sampler or equivalent	Immediately freeze with dry ice to below 0°C	7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Organic-free reagent water and ≤6°C	Laboratory freeze within 48 hours; analysis within 7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Na ₂ SO ₄ and ≤6°C	48 hours
		D: 5g soil plug added to 40 mL jar with septum cap	Methanol and ≤6°C	14 days
		MI: 1 L Glass Jar(*)	Methanol and ≤6°C	14 days
SVOCs	3550/8270	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	14 days
		D: Stainless Steel or Brass Sleeve		

Laboratory Methods, Containers, Preservation, and Holding Times for Soil Sample

Parameter	Method	Container	Preservation	Holding Time
		MIS: Dedicated Plastic Bag, 1 L Glass Jar(*)		
Metals (except Mercury and Chromium VI)	3050/6010 3050/6020	D: 4 oz Glass Jar with Teflon-Lined Cap	None	6 months
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
Mercury	7471	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	28 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
Chromium VI	7196	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 30 days A: 7 days
		D: Stainless Steel or Brass Sleeve in stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
PCBs PCBs	3540/8082 3550/8082 8081 (Aroclors Only)	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days A: 40 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
Chlorinated Herbicides	8151	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days A: 40 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
Organochlorine Pesticides	8081	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days A: 40 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
Triazine Pesticides	8141	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 14 days

Laboratory Methods, Containers, Preservation, and Holding Times for Soil Sample

Parameter	Method	Container	Preservation	Holding Time
		D: Stainless Steel or Brass Sleeve		A: 40 days
		MIS: Dedicated Plastic Bag		
Organophosphorus Pesticides	8141	D: 4 oz Glass Jar with Teflon-Lined Cap		E: 14 days
		D: Stainless Steel or Brass Sleeve	≤6°C	A: 40 days
		MIS: Dedicated Plastic Bag		
Carbamates	8321	D: 4 oz Glass Jar with Teflon-Lined Cap		E: 14 days
		D: Stainless Steel or Brass Sleeve	≤6°C	A: 40 days
		MIS: Dedicated Plastic Bag		
Fumigants	8260	D: EnCore® sampler or equivalent	≤6°C	48 hours
		D: EnCore® sampler or equivalent	Immediately freeze with dry ice to below 0°C	7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Organic-free reagent water and ≤6°C	Laboratory freeze within 48 hours; analysis within 7 days
		D: 5g soil plug added to 40 mL jar with septum cap	Na ₂ SO ₄ and ≤6°C	14 days
		D: 5g soil plug added to 40 mL jar with septum cap	Methanol and ≤6°C	14 days
		MI: 1 L Glass Jar(*)	Methanol and ≤6°C	14 days
Pentachlorophenol	8270 8151	D: 4 oz Glass Jar with Teflon-Lined Cap		E: 14 days
		D: Stainless Steel or Brass Sleeve	≤6°C	A: 40 days
		MIS: Dedicated Plastic Bag	Methanol and ≤6°C	14 days
Glyphosate	EPA 547	D: 4 oz Glass Jar with Teflon-Lined Cap		
		D: Stainless Steel or Brass Sleeve	≤6°C	14 days
		MIS: Dedicated Plastic Bag		

Laboratory Methods, Containers, Preservation, and Holding Times for Soil Sample

Parameter	Method	Container	Preservation	Holding Time
Cyanide	9013 EPA 335.3	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	14 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		
Dioxins/Furans	8290 8280	D: 4 oz Glass Jar with Teflon-Lined Cap	≤6°C	E: 30 days A: 45 days
		D: Stainless Steel or Brass Sleeve		
		MIS: Dedicated Plastic Bag		

Notes:

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- BTEX Benzene, toluene, ethylbenzene, and xylene
- Dioxins Polychlorinated dibenzodioxins
- Furans Polychlorinated dibenzofurans
- HVOCs Halogenated volatile organic compounds
- MTBE Methyl-tert butyl ether
- PAHs Polynuclear aromatic hydrocarbons
- PCBs Polychlorinated biphenyls
- SVOCs Semi-volatile organic compounds
- TPH-G Total petroleum hydrocarbons as gasoline
- TPH-D Total petroleum hydrocarbons as diesel
- TPH-O Total petroleum hydrocarbons as oil
- VOCs Volatile organic compounds
- ≤ Less than or equal to
- °C Degree Celsius
- g Gram
- L Liter
- mL Milliliter
- Na₂SO₄ Sodium bisulfate
- D Discrete
- MIS *Multi-Increment* sample
- E Hold time to extraction
- A Hold time after extraction until analysis
- (*) A methodology for the collection of *Multi-Increment* samples for volatile analysis has been proposed by the HEER Office (See [Section 4.2.7](#). The sample container must be large enough to accommodate 30 to 50 soil increments, preserved with 1 milliliter of methanol per 1 gram of soil, and additional methanol, as needed.
- EPA Where the term "EPA" is used with a given method number the prefix indicates that the method comes from Environmental Protection Agency (EPA) wastewater and drinking water standards, both published and maintained by the EPA Safe Drinking Water Act (SDWA) and/or Clean Water Act (CWA). For methods that are presented without the "EPA" notation, the methods come from the guidance document "Test Methods for

Evaluating Solid Waste Physical/Chemical Methods" (SW-846). Although SW-846 was written by the EPA originally, they are guidance documents and not prescriptive as the EPA prefix methods.



TGM for the Implementation of the Hawai'i State Contingency Plan
Appendix 11-B
Laboratory Methods for Groundwater

Appendix 11-B

Laboratory Methods, Containers, Preservation, and Holding Times for Groundwater Samples

Parameter	Method	Container	Preservation	Holding Time
TPH-G	5035/8015; LUFT	40 mL jar with septum cap	≤6°C	7 days
			HCl to pH<2 and ≤6°C	14 days
TPH-D	5035/8015; 3550/8270; 3540/8270; LUFT	1 L Amber Glass Jar	≤6°C	7 day (4°C only)
TPH-O	8015; EPA 1664	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
			HCl to pH<2 and ≤6°C (1664 only)	
BTEX, MtBE	(BTEX only) 8260; EPA 602; EPA 624	40 mL jar with septum cap	≤6°C	7 days
	(BTEX and MTBE) 8260; EPA 524.2		HCl to pH<2 and ≤6°C	14 days
PAHs	8270; 8310; EPA 610; EPA 625	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
HVOCs	8260; EPA 601; EPA 608; EPA 624; EPA 625	40 mL jar with septum cap	≤6°C	7 days
			HCl to pH<2 and ≤6°C	14 days
VOCs	8260; EPA 624	40 mL jar with septum cap	4°C	7 days
			HCl to pH<2 and ≤6°C	14 days
SVOCs	8270; EPA 625	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Metals (except Mercury and Chromium VI)	6010; 6020; EPA 200 Series	250 mL Plastic Container	HNO ₃ to pH<2 and ≤6°C	6 months
Mercury	7470; EPA 245.1	250 mL Plastic Container	HNO ₃ to pH<2 and ≤6°C	28 days
Chromium VI	7196	250 mL Plastic Container	≤6°C	24 hours
PCBs	8082; EPA 608	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Chlorinated Herbicides	8151; EPA 615	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Organochlorine Pesticides	8081; EPA 608	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days

Laboratory Methods, Containers, Preservation, and Holding Times for Groundwater Samples

Parameter	Method	Container	Preservation	Holding Time
Triazine Pesticides	8141; 8270; EPA 619	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Organophosphorus Pesticides	8141; 8270; EPA 622; EPA 614	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Carbamates	8321; EPA 632	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Fumigants	8260; EPA 504.1; EPA 524.2	40 mL jar with septum cap	≤6°C	7 days
			HCl to pH<2 and ≤6°C	14 days
Pentachlorophenol	8270; 8151; EPA 625	1 L Amber Glass Jar	≤6°C	E: 7 days A: 40 days
Glyphosate	EPA 547	1 L Amber Glass Jar	0.008% Na ₂ S ₂ O ₃ (pH 5-8) and ≤6°C	14 days
Cyanide	9014; EPA 335.2/335.3	500 mL Plastic Container	NaOH to pH>12 and ≤6°C	14 days
Dioxins/Furans	8290; EPA 613; EPA 1613	1 L Amber Glass Jar	≤6°C	E: 30 days A: 45 days

Notes:

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- BTEX Benzene, toluene, ethylbenzene, and xylene
- Dioxins Polychlorinated dibenzodioxins
- Furans Polychlorinated dibenzofurans
- HVOCs Halogenated volatile organic compounds
- MTBE Methyl-tert butyl ether
- PAHs Polynuclear aromatic hydrocarbons
- PCBs Polychlorinated biphenyls
- SVOCs Semi-volatile organic compounds
- TPH-G Total petroleum hydrocarbons as gasoline
- TPH-D Total petroleum hydrocarbons as diesel
- TPH-O Total petroleum hydrocarbons as oil
- VOCs Volatile organic compounds
- ≤ Less than or equal to
- °C Degree Celsius
- g Gram
- L Liter
- mL Milliliter
- HCl Hydrochloric acid
- NaOH Sodium hydroxide
- NaS₂O₃ Sodium thiosulfate
- HNO₃ Nitric acid
- E Hold time to extraction
- A Hold time after extraction until analysis
- pH A measure of the acidity or alkalinity of a solution.

EPA

Where the term "EPA" is used with a given method number the prefix indicates that the method comes from Environmental Protection Agency (EPA) wastewater and drinking water standards, both published and maintained by the EPA Safe Drinking Water Act (SDWA) and/or Clean Water Act (CWA). For methods that are presented without the "EPA" notation, the methods come from the guidance document "Test Methods for Evaluating Solid Waste Physical/Chemical Methods" (SW-846). Although SW-846 was written by the EPA originally, they are guidance documents and not prescriptive as the EPA prefix methods.